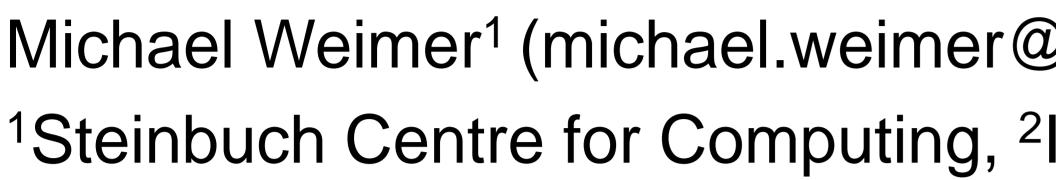
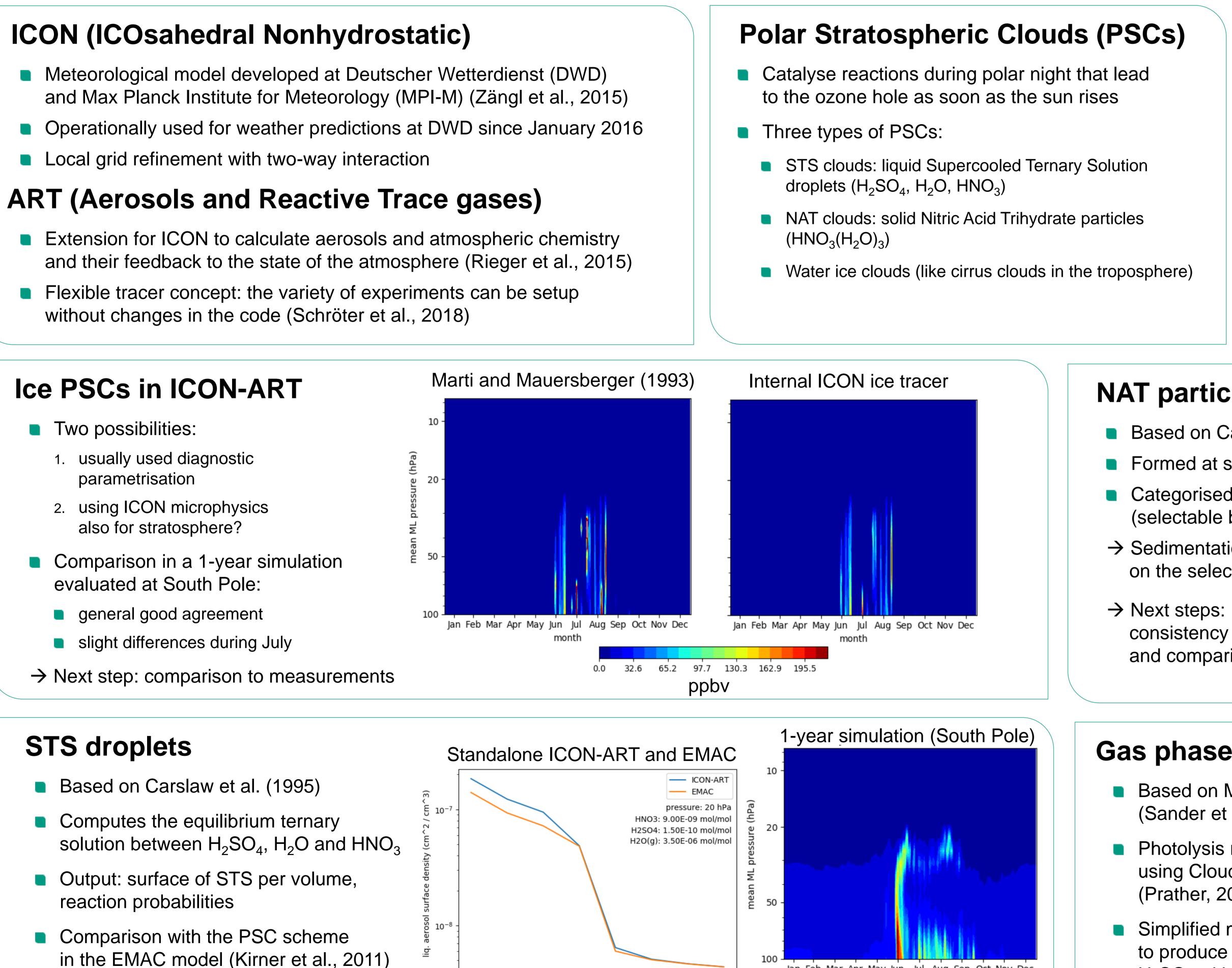


# **Polar Stratospheric Clouds in ICON-ART**



- without changes in the code (Schröter et al., 2018)



 $\rightarrow$  Next step: comparison to measurements



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Jan Feb Mar Apr May Jun Jul Aug Sep Oct Nov Dec

18.2

0.0

36.4 54.6 72.8 91.0 109.2

 $cm^2 / cm^3$ 

184

186

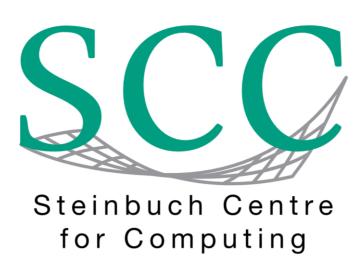
temperature (K)

178

190

192

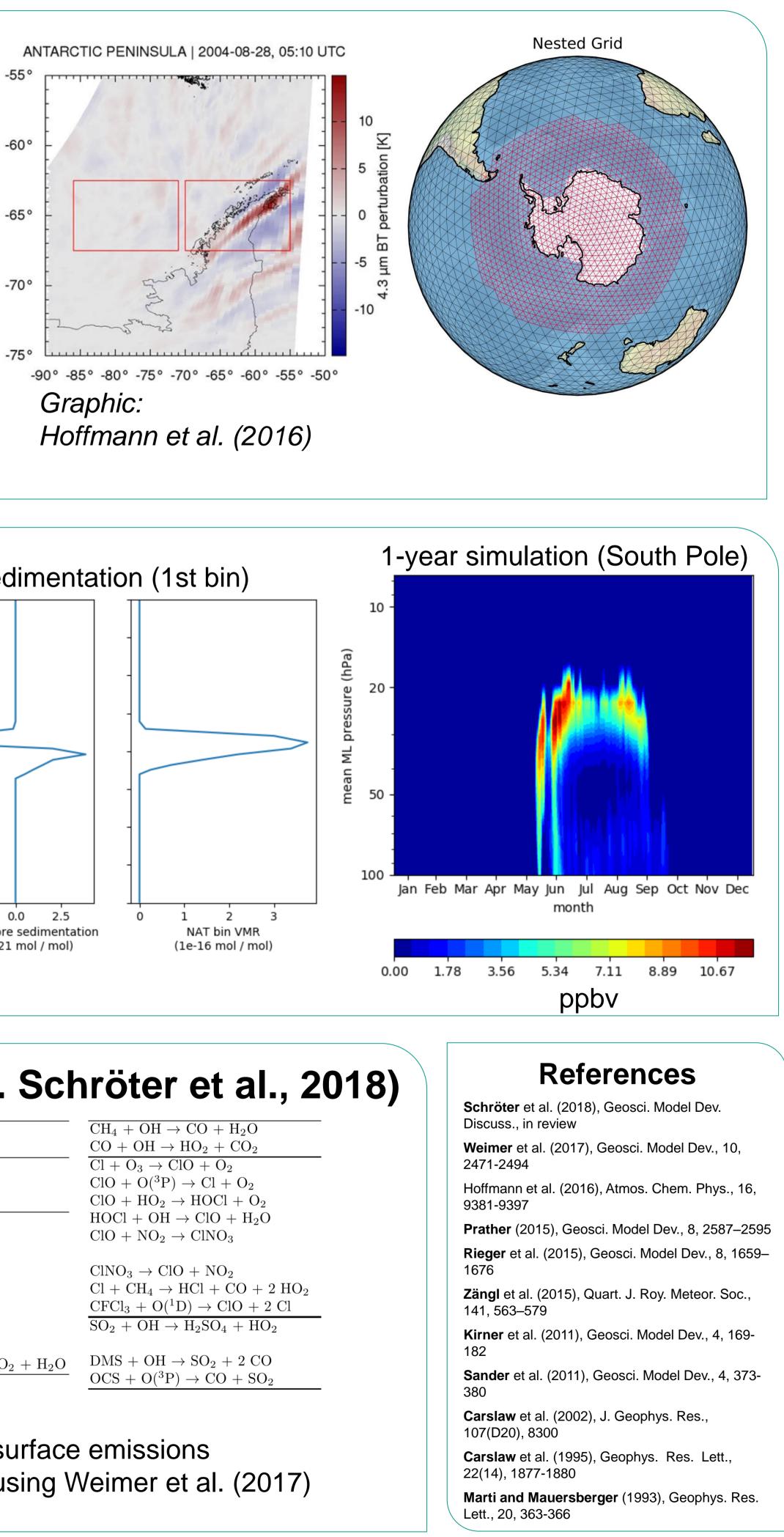
188



# Steinbuch Centre for Computing Simulation Laboratory Climate and Environment

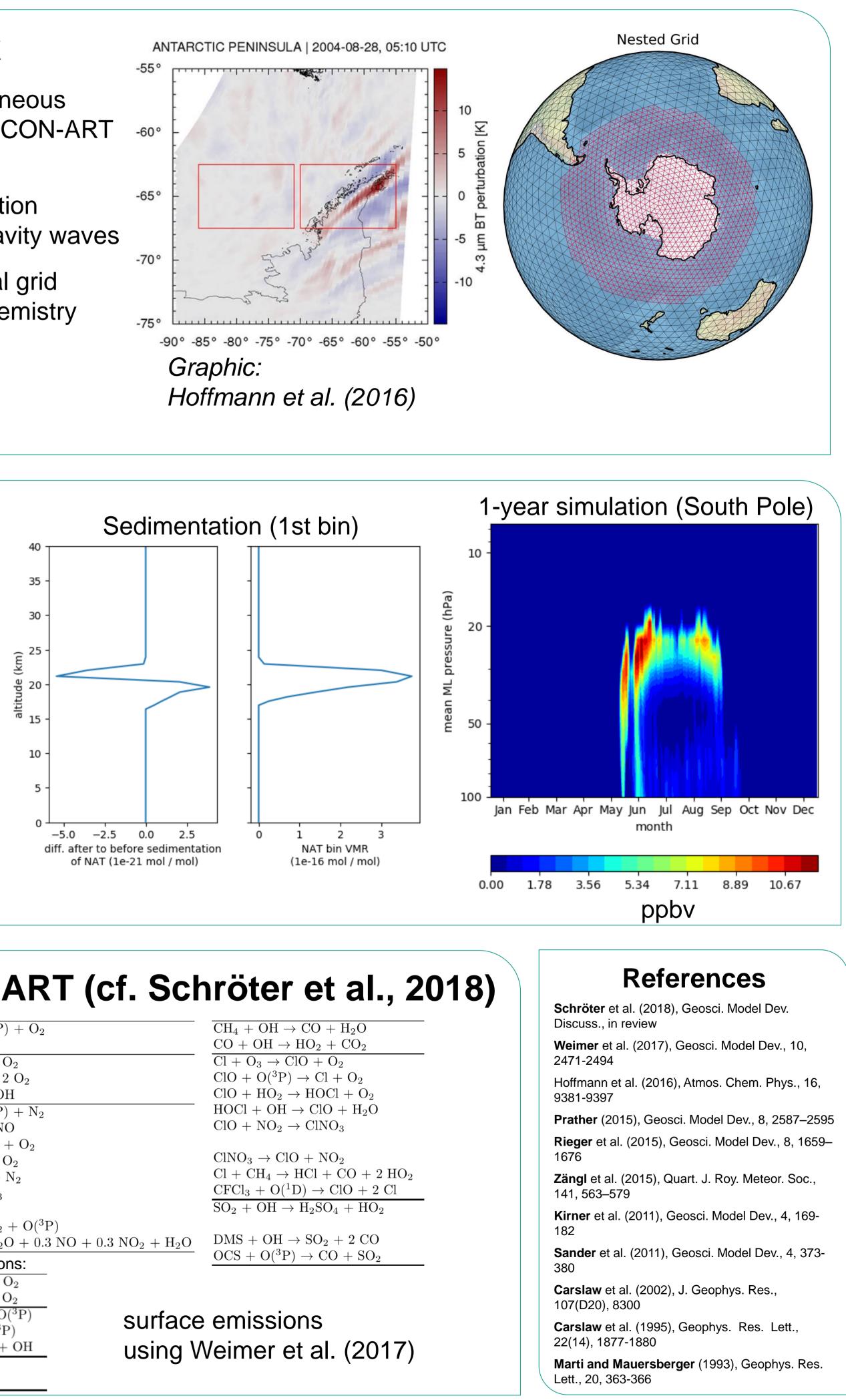
### Goal of this work

- Evaluation of heterogeneous chemistry on PSCs in ICON-ART -60° with satellite data
- Investigation of interaction between PSCs and gravity waves
- First application of local grid refinement (nest) in chemistry with ICON-ART
- Nested region around Antarctic Peninsula



# **NAT** particles

- Based on Carslaw et al. (2002)
- Formed at supersaturation
- Categorised in different size bins (selectable by user)
- $\rightarrow$  Sedimentation dependent on the selected size distribution
- consistency check of sedimentation and comparison to measurements



# Gas phase chemistry in ICON-ART (cf. Schröter et al., 2018)

- Based on MECCA (Sander et al., 2011)
- Photolysis rates using Cloud-J (Prather, 2015)
- Simplified mechanism to produce HNO<sub>3</sub> and H<sub>2</sub>SO<sub>4</sub> with ICON-ART
- Ozone depletion with simplified chlorine cycle

 $O_2 + O(^1D) \rightarrow O(^3P) + O_2$  $O_2 + O(^3P) \rightarrow O_3$  $OH + O_3 \rightarrow HO_2 + O_2$  $HO_2 + O_3 \rightarrow OH + 2 O_2$  $H_2O + O(^1D) \rightarrow 2 OH$  $\overline{N_2 + O(^1D)} \rightarrow O(^3P) + N_2$  $N_2O + O(^1D) \rightarrow 2 NO$  $N_2O + O(^1D) \rightarrow N_2 + O_2$  $NO + O_3 \rightarrow NO_2 + O_2$  $NO + N \rightarrow O(^{3}P) + N_{2}$  $NO_2 + OH \rightarrow HNO_3$ 

 $HNO_3 + OH \rightarrow NO_2 + O(^3P)$  $NH_3 + OH \rightarrow 0.2 N_2O + 0.3 NO + 0.3 NO_2 + H_2O$ photolytic reactions:  $O_3 + h\nu \rightarrow O(^1D) + O_2$  $O_3 + h\nu \rightarrow O(^3P) + O_2$ 

 $NO_2 + h\nu \rightarrow NO + O(^3P)$  $NO + h\nu \rightarrow N + O(^{3}P)$  $\frac{\mathrm{HNO}_3 + \mathrm{h}\nu \rightarrow \mathrm{NO}_2 + \mathrm{OH}}{\mathrm{Cl}_2 + \mathrm{h}\nu \rightarrow \mathrm{Cl} + \mathrm{Cl}}$  $CFCl_3 + h\nu \rightarrow 3 Cl$ 

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